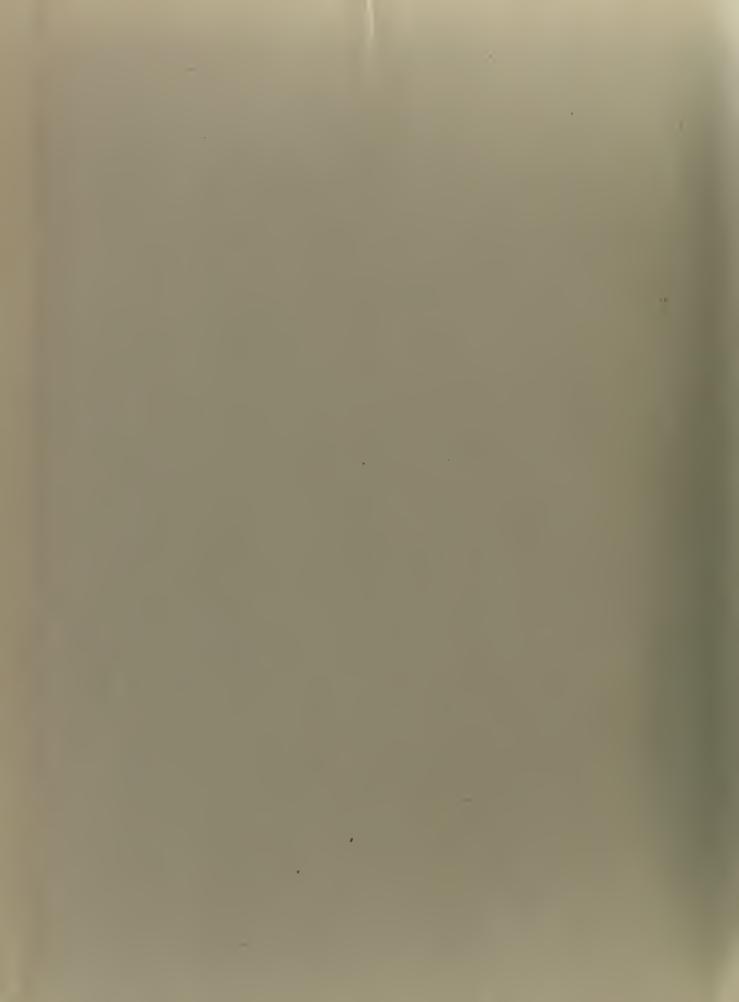
FEASBLIT OF A COMPACTAXIAL FLOW NUCLAR HEAT EXCHANGE

Charles Anthony Kiser













PEASIBILITY OF A COMPACT AXIAL PLOW NUCLEAR HEAT EXCHANGER

by

Charles Anthony Kiser

A Thesis Submitted to the
Graduate Faculty in Partial Fulfillment of
The Requirements for the Degree of
MASTER OF SCIENCE

Major Subject: Engineering

Approved:

7/4

CONTRACTOR OF THE PERSON NAMED IN COLUMN 2 IN COLUMN 2

CHARLES AND ADDRESS OF THE PARTY OF

the second section of the section of

and the second second

Approved

TABLE OF CONTENTS

| | Page |
|-----------------------------------|----------------|
| ABSTRACT | iii |
| INTRODUCTION | 1 |
| LIST OF SYMBOLS | 3 |
| LIMITING CONDITIONS | 7 |
| HEAT TRANSFER ANALYSIS | 10 |
| Development Discussion Conclusion | 10 17 20 |
| FUEL ELEMENT CONSIDERATION | 21 |
| Development Discussion Conclusion | 21 22 26 |
| NUCLEAR ANALYSIS | 27 |
| Development Discussion Conclusion | 27 30 33 |
| OVER-ALL FEASIBILITY CONCLUSIONS | 35 |
| LITERATURE CITED | th. |
| APPENDIX A | 46 |
| APPENDIX B | 49 |
| APPENDIX C | 54 |
| APPENDTY D | K8 |

ABSTRACT

computations indicate that when utilized in an idealized ramjet at a flight Mach number of 2.5 and an altitude of 37,500 feet, a constant area axial flow nuclear heat exchanger will be expected to produce 9,500 pounds of thrust with an air inlet Mach number of 0.2, a weighted radial temperature average of 2000°P and a frontal area of 7.7 square feet. Molybdenum alloy with a disilicide coating was selected as the container material. Sandwiched between the molybdenum was a fissionable alloy of highly enriched uranium and zirconium. From a nuclear aspect the unreflected heat exchanger becomes critical at a diameter of 40 inches and a length of 60 inches with a total weight of 8,052 pounds which includes 3,622 pounds of 90 per cent enriched uranium.

THE OWNER OF THE OWNER, THE OWNER,

conversables followed to be seen to be and the last blood and the last

INTRODUCTION

That a constant-area axial-flow compact nuclear heat exchanger as envisioned in Figure 1 when installed in an idealized ramjet would produce significant thrust was the basis of this thesis. Feasibility conclusions were limited to the information revealed by heat transfer and nuclear analysis under one assumed steady state condition. The problems of nuclear control, shielding, thermal stresses, and performance under different inlet Mach numbers were not considered.

The nuclear heat exchanger as shown in Figure 2 incorporates thin concentric cylinders which are the solid fissionable containers. The alloyed fissionable phase is sandwiched between two layers of non-fissionable structural material so that the enriched uranium fuel cores are surrounded by a continuous network of structural metal for strength and containment at high temperatures. Air flows within the annuli and acts as the coolant and thrust agent. It is recommended that the nuclear flux be flattened by a central control core and a peripheral reflector; however, the beneficial effects of a reflector and central control core were not considered.

From the assumed inlet condition, various maximum cylinder or wall temperatures were selected arbitrarily for
thrust results. That the high temperature fuel elements and

-

The same of the sa

THE RESIDENCE AND THE PARTY AND PERSONS AND PARTY AND PA

structural materials were possible metallurgically were next investigated. Whether or not an unreflected homogeneous reactor of this configuration can be made critical was determined by using one group diffusion theory.

And the state of the control of the

LIST OF SYMBOLS

```
Cross section area normal to flow, (ft2), or the mass
            number of an element (gm/gm atomic weight).
       Wetted area, (ft2).
A
       Buckling. (cm<sup>-2</sup>).
B
       Local velocity of sound, (ft/sec).
C
       Specific heat at constant pressure, (BTU/16 °F).
C
       Diffusion coefficient for flux, (cm).
D
       Hydraulic diameter, Dp - D1, (ft).
D
DI
       Inside diameter, (ft).
       Outside diameter, (ft).
Da
       Neutron leakage, (n/cm3 sec).
-DV2d
       Friction factor defined as 25 T / pv2.
f
       Acceleration of gravity (32.2 ft/sec2).
8
       Mass velocity (lb/sec ft2).
G
       Coefficient of heat transfer between fluid and surface.
h
             (BTU/hr ft2 of).
```

he based on (To,w - To,aw).

h based on logarithm mean of (To.w - T).

h based on (Tow - T).

hs based on (To,w - To).

H Extrapolated distance, (cm).

k Neutron multiplication factor.

K Thermal conductivity, (BTU/hr ft2 of/ft).

AMERICA DE SEASON

- AND AND HE SPECIAL PROPERTY AND ADDRESS OF THE PARTY ADDRESS OF THE PART
 - of the same name of
 - A THE PERSON NAMED IN
 - promoted to the second to the second
 - ALT ANTO A CONTRACT OF THE PARTY OF THE PART
 - which you'll not describe the otherwise.
 - with the same special and the same
 - OFFI CHICAGO STATE OF
 - The water manager of the
 - A TANK A STREET, MARRIED BY STRE
 - Non-American and American St.
 - A CONTRACTOR OF STREET
 - the second course are
- AND THE RESIDENCE OF THE PARTY OF THE PARTY
 - Country of the latest and
 - All the second of the last of
 - all to apply the formal gar
 - 10 10 m hand of
 - THE PERSON NAMED IN
 - Annual or the Park State of the Owner, where the
 - STATE OF STREET STATES AND ADDRESS OF THE PARTY NAMED IN

```
Length of heat exchanger heating element in the
L
             longitudinal direction, (ft).
       Mass rate of flow, (lb/sec).
m
       Mach number or molecular weight (gms/gm molecular
M
             weight).
       Number of neutrons per cm3.
n
       Number of nuclei per em3.
M
       Avogadros number (6.02 x 10<sup>23</sup> nuclei/gm atomic weight).
No
       Prandtl number, C_\mu/K.
NPR
       Reynolds number, evD / \mu.
NBE
       Stanton number, he/CnG.
NST
       Absolute stream pressure (lb/in2).
P
        Isentropic stagnation pressure (lb/in2).
Po
       Heat-transfer rate (BTU/hr).
q
       Volumetric heat source which is equal to E E, p,
Q
           (BTU/sec cm<sup>3</sup>).
       Radius (ft).
       Absolute stream temperature (°R).
T
Th
       Thrust (1b).
To
       Stagnation temperature (OR).
To,aw
       Adiabatic stagnation wall temperature ( PR).
To, w
       Stagnation wall temperature (°R).
       Velocity (ft/sec).
       Volume (ft3 or cm3).
V
```

Distance along heat exchanger heating elements (ft).

W

X

Weight (1b).

- AND AND STREET, SPECIAL PROPERTY OF THE PARTY OF
- A desirable part to our and de
 - malaurahan anglyangsi ringston restricted on resistant about 10
 - . For your providence has recorded in
 - Asia and restrain to beauty to
- of Region Administration of Contract of the Park of Street, Square and Square
 - Aller or the same of the same
 - THE PERSON NAMED IN
 - Mary reduce sectors, og 17, 51.
 - a Challette and the state of the late of t
 - CANADA AND THE PARTY W.
 - Against him of distributions but terminal 9
 - . .
 - .101 miles 8
 - A 1971 THE RESIDENCE PROPERTY AND ADDRESS OF
 - a Street (Es)
 - The statement of the later of
 - They between the otherwise of the ball of
 - (11) summer of the self-real top
 - A District Whitehalls
 - . The select mater a
 - VISTO CONTRACT NO.
 - which amounts making transmit that body would be

Subscripts

b Bulk (T).

f Pilm $t_f = (t_b + t_w)/2$.

w Wall.

0,1,2 Station references

Greek

a Definition of mx/L.

8 Ratio of specific heats of gas, $C_p/C_v = 1.4$.

Ah Change in enthalpy, (BTU/1b).

7 Number of neutrons produced per absorption in the fissionable isotope.

 λ Definition of $(N_{PR})_f^{-2/3}$.

λ transport mean free path, (cm).

Absolute viscosity of the fluid, (lb/sec ft).

Average cosine of the scattering angle per neutron collision in the laboratory system.

Mumber of neutrons produced per fission in the fissionable isotops.

ρ Density, (lb/rt³).

o Microscopie cross section (cm²).

 σ_a microscopic absorption cross section (cm²).

 σ_n microscopic scattering cross section (cm²).

 $\sigma_{\rm f}$ microscopic fission cross section (cm²).

of the state of th

 Σ Macroscopic cross section (cm⁻¹), (N σ).

 $\Sigma_{\rm s}$ macroscopic absorption cross section (cm⁻¹).

Σ_s macroscopic scattering cross section (cm⁻¹).

Σ, macroscopic fission cross section (cm⁻¹).

 Σ_{n} Number of neutron absorptions / cm³ sec.

 T_w Tractive force at wall due to fluid friction (lb/ft²).

Flux in neutrons / cm² sec.

V2 Laplacian operator.

Phil interes are alternated attended in the control of the control

· - - - -

and the second second and the second second

LIMITING CONDITIONS

To obtain a realistic operating condition for the aircooled nuclear heat exchanger the following conditions were assumed:

$$M_0 = 2.5$$
, $h = 37,500 \text{ ft}$, $T = 392°R$, $P = 3.09 \text{ lb/in}^2$

Godsey and Young (1, p.221) stated that the total temperature through the inlet diffuser depends only on flight Mach number and ambient temperature and is a constant regardless of diffuser losses if the process is considered adiabatic.

Therefore the relation

$$T_{0,0} = T_{0,1} = T_{0,2} = T(1 + \frac{\gamma - 1}{2} M_0^2)$$
 (1)

may be used where the station numbers correspond to those of Figure 1. Godsey and Young (1, p.221) stated that the total pressure at the heat exchanger inlet, $P_{0,2}$, was dependent upon the flight Mach number, ambient pressure, free stream diffusion losses, duct friction losses, and duct diffusion losses. The last three effects may be grouped under the term ram effectiveness or pressure recovery effectiveness. From data presented by Godsey and Young (1, p.86), the ram effectiveness at $M_0 = 2.5$ was given as approximately 0.98; however, subsonic diffusion was assumed to be loss free. For a more realistic condition a value of 0.90 was chosen. The above ratios may be

The same of the sa

written out in the following expression for stagnation pressure at the heat exchanger inlet:

$$P_{0,2} = P_0(1 + \frac{Y-1}{2} M_0^2) / r-1 \times \frac{P_{0,2 \text{ actual}}}{P_{0,2 \text{ ideal}}}$$
 (la)

Solutions to Equations 1 and la give $T_{0,2} = 884$ °R and $P_{0,2} = 46.5$ lb per sq in. for inlet conditions to the heat exchanger.

Prior to converting the above inlet stagnation pressure to stream pressure, it is necessary to examine the limiting conditions for heat exchange. Assuming a uniform cross sectional heat exchanger area under a given inlet stagnation temperature, which in this case is 884°R, and various inlet Mach numbers, it is desired to exchange heat to the extent that the stagnation temperature at the exit of the heat exchanger be raised to a maximum value. By use of methods outlined in Reference (2, p.4.24) the results of various inlet Mach numbers are shown on Figure 3 which reveal that when a certain maximum temperature occurs, thermal choking even under frictionless conditions exists in the heating process and definitely limits the capacity of the constant area combustor or heat exchanger. Stated another way, if at any position downstream for a constant area duct, the stagnation temperature is equal to To.3 maximum, the flow is choked since the Mach number is unity. Any addition of friction

THE RESERVE AND ADDRESS OF THE PARTY OF THE

the state of the same of the s

market on the same of particular of the particular distance of NAME AND ADDRESS OF TAXABLE PARTY AND POST OFFICE ADDRESS. DESCRIPTION AND ADDRESS OF THE OWNER OF THE PERSON NAMED IN COLUMN 2 IN COLUMN DATE OF THE PARTY the first own to the second of the party of THE RESERVE TO SECURE ASSESSMENT OF THE PARTY OF THE PART the second of the second or the parties of regions. manufacture of the second seco AND REAL PROPERTY AND ADDRESS OF THE PARTY AND and the second s the same of the party of the pa managed in madellike upon against all makes along their and annual

will reduce the downstream interval for To,3 maximum to occur. With M₂ = 0.2 and To,3 maximum achieved, 5100°R, the stagnation pressure would suffer a 20 per cent loss due entirely to the heating effect.

Without becoming involved with a specific shock cone and diffuser design, the inlet Mach number was set at 0.2, which resulted in the following stream conditions:

As shown by Figure 2, the heat exchanger elements are thin hollow constant concentric area right cylinders which are adaptable to one-dimensional compressible gas flow treatment without introducing serious errors. This assumes that in any one annulus the change of stream properties in the direction of flow is much larger than in the radial direction.

the factor of the formation of the first terms of the state of the sta

AND REST OFFICE OF THE REAL PROPERTY AND ADDRESS OF THE PARTY ADDRESS OF THE PARTY AND ADDRESS OF THE PARTY ADDRESS OF THE

AND DESCRIPTION OF THE PARTY OF

,

HEAT TRANSFER ANALYSIS

McAdams (3, p.289) correlated the experimental data of Kays and London (4) and because compactness of an air-borne heat exchanger is an important parameter, a comparison of heat transfer coefficients per unit volume for a given friction power per unit volume showed ruffled fins and louvered plate fins as being the most effective. It appears that the highest ratio of heat transfer to pumping power is obtained when form drag is absent, or when all the drag is due to skin friction which is the case in common for flat plates or fins. Even though externally finned tubes are of value in increasing the rate of heat transfer, friction losses become prohibitive for this type of installation. For these reasons thin concentric hollow cylinders were chosen as the fuel elements of the heat exchanger and were restricted to constant longitudinal cross section area for nuclear simplification.

Development

If heat loss and longitudinal thermal conductivity are neglected, the rise in the enthalpy of the air per second is related to the effective coefficient of heat transfer from an annulus by a differential equation which is developed in Appendix C as

affection or on our name

The control of the control of the control of the state of the state of the control of the contro

.....

The same and the same in particular and the same and and and and an anticular anticular anticular and an anticular and an anticular and an anticular anticula

$$\frac{\pi}{h} (D_2^2 - D_1^2) \rho v C_p dT_0 = h_e \pi (D_2 + D_1) (T_{0,w} - T_{0,aw}) dx.$$
 (2)

The particular temperature differential is shown schematically in Figure 4. The temperature increment for heat transfer is the difference between the actual wall temperature, $T_{0,w}$, and the temperature $T_{0,aw}$, which the wall would assume for the same flow conditions but with zero heat flux. McAdams et al. (5) found that the effective heat transfer coefficient, h_e , expressed in terms of the Stanton number, N_{ST} , was independent of the temperature potential, $T_{0,w} - T_{0}$, when based on $T_{0,w} - T_{0,aw}$ while the coefficients h_m and h_s were not independent.

The Reynolds analogy with $N_{PR} = 1$ relates the Stanton number to the friction factor by

$$\frac{h_e}{C_p G} = \frac{f}{2} \qquad (3)$$

From experimental data for a moderate ΔT , $L/D_e = 60$, and $5,000 < N_{RE} < 200,000$, the following equations from McAdams (3, pp.155,219) may be combined

$$f = .046/(DG/\mu_f)^{.2} \tag{4}$$

$$\frac{h_{L}}{c_{pb}G} \left(\frac{c_{p}\mu}{K}\right)_{f}^{2/3} = \frac{.023}{(DG/\mu_{f})^{.2}}$$
 (5)

THE RESIDENCE OF THE PERSON NAMED IN

to produce

$$\frac{h_L}{c_{pb}G} = \frac{\lambda f}{2} \tag{6}$$

where, by definition, $\lambda \equiv (N_{\rm PR})_{\rm f}^{-2/3}$. Since Equations 3 and 6 are essentially the same, the limitation to a moderate temperature change may be removed without introducing serious errors when using the previously defined temperature potential. The following relationship, where specific heat is considered a constant, is used for the remainder of the analysis:

$$\frac{h_0}{C_p G} = \frac{\lambda f}{2} \tag{7}$$

The friction factor of Equation 7 is set by Equation 4 which is dependent upon Reynolds number and the degree of roughness. Since mass velocity, 6, and specific heat are essentially constant, the friction factor may be considered to depend upon hydraulic diameter, viscosity, and degree of roughness. An average value of $\lambda_1 = .008$ was obtained from Equation 4 for $1/2^n < D_e < 1-1/2^n$ and $1400^{\circ}R < \mu_f < 1600^{\circ}R$. The variance of the friction factor within the above range of hydraulic diameter may be compensated for by added degrees of roughness; however, Cope (6) found that with water in the turbulent region even though friction was six times that for smooth pipes, the heat transfer was only 20 to 100 per cent greater than for smooth tubes. McAdams (3, p.223) states that because of a very

A few of contrasts of the property of the contrast of the property of the contrast of the property of the contrast of the cont

small temperature change the heat transfer coefficients obtained were somewhat uncertain. Correlation with air may not apply due to different Prandtl number effect; however, there is a possibility that with small variation in hydraulic diameter the friction factor could remain a constant by varying the degrees of roughness.

Experimental data have shown that for turbulent flow in rectangular ducts and annular spaces friction may be evaluated from data for circular pipes using a hydraulic diameter shown in Figure 2 as

$$D_e = \mu_h = \frac{\pi(D_2^2 - D_1^2)}{\pi(D_2 + D_1)} = D_2 - D_1$$
 (8)

A combination of Equations 2 and 7 with Equation 8 results in

$$\frac{d T_0}{T_0} = \frac{T_{0,w}}{T_0} - \frac{T_{0,aw}}{T_0} = 2 \lambda f \frac{dx}{D_e} . \qquad (9)$$

If a gas stream of uniform temperature is brought to rest isentropically, as at the stagnation of a blunt body, the temperature rise for an ideal gas will be

$$T_0 - T = \frac{V^2}{2g JC_p} = T(1 + \frac{8-1}{2} M^2)$$
 (10)

For convenience a temperature recovery factor N_{RF} as shown in Figure 4 is defined as

AND DESCRIPTION WHEN AND ADDRESS OF PERSONS ASSESSED TO THE PARTY OF T

the next parameter our root room man and independent because on an independent many rollings are rised velocities many regional of the parameter and a rolling and a series and a as a result of

all managed it makes all over it was a proportioned by encountered by

the second of the second received product of the second state of t

$$N_{RF} = \frac{T_{0,aw} - T}{T_{0} - T} = \frac{T_{0,aw} - T}{v^2/2g JC_p}$$
 (11)

By a combination of Equation 10 and 11, the stream temperature, T, may be eliminated to produce

$$\frac{T_{0,aw}}{T_{0}} = 1 - \frac{(1 - N_{RF}) \frac{y-1}{2} m^{2}}{1 + \frac{y-1}{2} m^{2}}.$$
 (12)

Substitution of Equation 12 into 9, yields

$$\frac{dT_0}{T_0} = \frac{T_{0,W}}{T_0} - 1 + \frac{(1 - N_{RP}) \frac{r-1}{2} M^2}{1 + \frac{r-1}{2} M^2} = 2 \lambda f \frac{dx}{D_0} . \quad (13)$$

The generalized differential Mach number equation of Reference (2, p.4.12) simplifies to

$$\frac{dM^{2}}{M^{2}} = \frac{(1 + 8 M^{2})(1 + \frac{8-1}{2} M^{2})}{\frac{dT_{0}}{T_{0}}} + \frac{M^{2}(1 + \frac{8-1}{2} M^{2})}{1 - M^{2}} + \frac{dx}{D_{e}}$$
(14)

with negligible structure drag forces and constant specific heat, molecular weight, area and mass flow rate.

The same of the sa

Trible J. sent to military be related to the

of malkaligness (things only assessment)

articles from the court has been proposed assessment and the same of the same

Equation 13 was substituted into Equation 14 which gave

$$dM^{2} = \frac{M^{2}(1 + \frac{\gamma - 1}{2} M^{2})}{1 - M^{2}} 2r \frac{dx}{D_{e}}.$$

$$\left\{ (1 + \delta M^{2})\lambda \left[\frac{T_{0,W}}{T_{0}} - 1 + \frac{1 - N_{RF} \frac{\gamma - 1}{2} M^{2}}{1 + \frac{\gamma - 1}{2} M^{2}} \right] + 2r M^{2} \right\}$$
(15)

For a turbulent boundary layer over a flat plate,
McAdams (3, p.311) recommends that the temperature recovery
factor and the Prandtl number be related by

$$N_{RF} = N_{PR} 1/3$$
 (16)

McAdams et al. (5) reported recovery factors for subsonic flow of air near the outlet and of a long smooth tube were in good agreement with those predicted by Equation 16. Appendix A shows the procedure for solving Equations 13 and 15 for one Mach number and stagnation temperature at an assumed position downstream. From these calculations it was evident that the contribution of the recovery factor term N_{RF}, was negligible for Mach numbers up to approximately 0.8; however, the resulting stagnation temperatures will be nearly 1.5 per cent higher than those which have incorporated this term.

For the remainder of this analysis the recovery factor and Prandtl number were assumed unity which reduced Equation 13 to

THE RESERVE AND ADDRESS OF PERSONS ASSESSED.

principal and the second secon

manufacture and a large state of the same and the same an

$$\frac{dT_0}{T_0} = \begin{bmatrix} \frac{T_{0,W}}{T_0} - 1 \end{bmatrix} 2f \frac{dx}{D_0}$$
 (17)

It is of interest to note that Equation 17 could be expected from experimental data by McAdams (3, p.314) because the heat transfer coefficient, h, decreases and approaches the constant value of h for temperature differences greater than 100°F. For a high temperature potential, h, approaches h and may be used directly in Equation 1.

Equations 14 and 17 were integrated approximately over short intervals of Mach numbers by using the procedure outlined in Reference (2, 4.60). Appendix B contains the stepwise procedure used and Tables 1-3, Appendix B, have the complete data for stagnation and stream properties as a function of L/D_e ratios for various longitudinally constant wall temperatures. Stagnation temperatures and pressures versus L/D_e ratios are shown in Figure 5.

However, from a nuclear standpoint the longitudinal flux of fuel element cylinders will follow a sine distribution with the maximum flux produced at the midpoint. This does not imply that the maximum wall temperature will occur at the longitudinal midpoint. By use of

$$T_{0,3} - T_{0,2} = \frac{2(T_{0,W \text{ max}} - T_{0,2})}{1 - \sec \alpha \text{ max}}$$
 (18)

The test of the party of the second of the s

The second of the control of the con

AND DESCRIPTION OF THE PERSON NAMED IN COLUMN TWO IS NOT THE PERSON NAMED IN COLUMN TO PERSON NAMED IN COLUMN TO

which was developed in Appendix C and by using the previously selected temperatures for maximum wall temperatures, the stagnation temperatures versus L/D ratios were computed as shown in Table 4 of Appendix C and plotted in Figure 5. Stagnation pressures for variable wall temperatures were obtained by interpolation from previously calculated pressures.

An isentropic expansion using air data from Keenan and Kaye (7) was assumed in order to obtain the change of enthalpy from stage 3 to 4 of Figure 1. The Brayton cycle as shown in Figure 6 represents an overall view; however, at various L/De ratios, the change of enthalpy was computed using an ambient pressure of 3.09 lb per sq in. Figure 7 shows the variance of enthalpy for both constant and variable wall temperatures. The right hand ordinate of Figure 7 related change of enthalpy to pounds of thrust per sq ft by

$$\frac{T_h}{A} = \frac{G}{g} \left(v_{l_{\downarrow}} - v_{Q} \right) \qquad , \tag{19}$$

where
$$v_0 = 2.5c$$
 and $v_{\downarrow} = 223.7\sqrt{h_{0,3} - h_{\downarrow}}$.

Discussion

The calculated values of stagnation temperatures and pressure as shown in Figure 5 do not show the contributing effects on heat transfer and friction caused by the length

THE RESIDENCE AND DESCRIPTION OF PERSONS AND PARTY AND P

The second and the second of t

The Administration of the state of the state of

ALCOHOLD !

the state of the same of the s

to diameter ratio and the effects of sudden enlargement or contraction of flow. The former effect is small since the L/D_e ratio for this heat exchanger should be greater than 60; the latter effect may still be appreciable even though the peripheral thickness of the cylinders are in the order of 0.05 to 0.07 in. The specific heat ratio was set at 1.4, and the friction factor, specific heat, and mass velocity were constant throughout the heating process. The cosine flux distribution in the longitudinal direction is justifiable for a finite cylinder.

The temperatures obtained by normal forced convection methods for an assumed T_{0,w} = 2460°R are shown in Figure 5 for comparison. The temperatures show good correlation with stagnation temperatures obtained by the Reynolds analogy but are considerably higher than the stream temperatures. The forced convection formulas will not show any decreasing trend in stream temperatures or a maximum stagnation temperature as heat is continually added as will the Reynolds method for compressible fluid which will eventually produce choked flow at a certain L/D_e ratio downstream.

Figure 7 shows that the effect on thrust reduction was quite severe for a 2000°F constant wall temperature versus a 2000°F maximum wall temperature; the thrust per square foot was reduced by 18 per cent. For a uniform velocity at the exit of stage 4, a constant change in enthalpy is required. If the neutron flux was such as to produce a series of maximum

A Desirable recognished the property of the pr

wall temperatures from 1600 to 2200°F and with a constant enthalpy change of 200 BTU per 1b imposed, the L/D_e ratio would vary from 47 to 100. Since all cylinders are desired to have the same length for nuclear considerations, the hydraulic diameters would vary by 47 per cent with decreasing hydraulic diameters as radial distance is increased. This diameter variance would still be subject to the limitation imposed on the hydraulic diameter due to a constant friction factor.

changer cross-sectional area, each fuel cylinder should operate at the maximum enthalpy change as shown by Figure 7 and the thrust would be a weighted average depending on the frontal area of the heat exchanger. With a constant length assumed the hydraulic diameter would vary by only 12 per cent. However, each annulus would be operating at a different velocity level which may create internal vibration within the ramjet pod. With a weighted radial average of 2000°F for the cylinder temperatures encompassing 7.7 sq ft of heat exchanger cross-sectional area, from Figure 7 the resultant thrust would be approximately 9,500 lb at a flight Mach number of 2.5 at 37,500 ft.

With inlet Mach numbers greater than 0.2, the L/D_e ratio would be substantially reduced. From Reference (2, p.431), with $M_2 = 0.5$, f = 0.005, $T_{0,2} = 1000^{\circ}R$, $P_{0,2} = 10 \text{ lb/in}^2$, and $T_{0,w} = 2000^{\circ}R$ and $1500^{\circ}R$, choked flow occurred at an L/D_e ratio

The second states of the second states and t

A STATE OF THE PARTY OF THE PAR

of 20 and 35, respectively. To resist the pancaking effect which may reach the limit for nuclear criticality, the inlet Mach number appears to have a definite upper limit when heating is desired.

Conclusion

For the assumed inlet conditions, cylindrical fuel elements and wall temperatures, the amount of heat transferred
was substantial enough to produce 1,000 to 1,400 pounds of
thrust per sq ft if heat exchanger frontal area with a maximum
wall temperatures range from 1600 to 2200°P at a flight Mach
number of 2.5 and an altitude of 37,500 ft. With a weighted
radial average of 2000°P for the cylinder temperatures encompassing 7.7 sq ft of heat exchanger area the resultant thrust
was 9,500 lb.

of the contract of a property of the party of the contract of

SECRETARIST STREET

An extraction of the former and process of the last of the same and th

FUEL ELEMENT CONSIDERATION

Due to the absence of effective moderation within this nuclear heat exchanger, the unit may be considered as a fast neutron reactor and as such the restrictions normally imposed on structural materials by high thermal neutron absorption may be removed. Whether any structural material now in existence can be used to contain, protect, and support the fissionable phase for the high temperatures required is the subject of the following development and discussion.

Development

The nuclear heat exchanger as shown in Figure 2 incorporates thin concentric cylinders which are the solid fissionable containers. The enriched uranium fuel cores are to remain a solid to prevent serious containment problems. The thin sheets of the fissionable phase are sandwiched between the inner and outer layers of the non-fissionable structural metal which predominates in volume so that the alloyed enriched uranium fuel sheets are surrounded by a continuous network of structural metal for strength at high temperatures.

The following are some criteria that were considered for the selection of fuel elements and structural material:

BUTTO DE TENDO DE

The control of the co

Annual Property lies and

The property of the second of the property of the second o

LANDSHAM LANDSHIPS for promote both to converse and

A. Fuel element criteria

- 1. Helting point above 2500°F
- 2. Neutron's energy maintained
- 3. Mechanical properties similar to the container material during fabrication and high temperature operation.
- 4. High thermal conductivity and low specific heat for good heat transfer
- 5. Ease of fabrication

B. Container material criteria

- 1. High temperature mechanical properties under static and small radial dynamic loads
- 2. Resistance to oxidation at high temperatures
- 3. High resistance to thermal shock
- 4. Neutron energy maintained
- 5. Ease of fabrication, rolling, and weldability.

Discussion

The melting point of natural uranium is 2072°F; for the required temperature range of 2000 to 2200°F, the unalloyed uranium would become a liquid and present serious containment problems. In order to increase the melting point, the various binary diagrams of uranium alloys of Saller (8, p.413-414) revealed that a uranium-zirconium system of 20.7 per cent zirconium by atomic weight would have a melting point of 2450°F.

- AND PERSONAL PROPERTY AND PERSONAL PROPERTY
 - The Co. oppose thing unitalist at
- Conduction System Branches Ch.
- -analytims and of margins addressed distributed of -analytims and the margins of the same distributed and the same distri
 - TO SHARE THE PARTY OF THE PARTY

- MANYOU SAFEVE SAMOTHER OF
- AND DESCRIPTION AND DESCRIPTION OF THE PARTY OF THE PARTY
- The second or management of
- moderation process control of
- the same of femalescent, well the own of

and the state of t

the part printing or a printing to the same of the sam

Elements with less than a mass number of 40 were not considered because of the increased slowing down effect on the neutrons.

Zirconium alloys at high temperatures suffer a loss of strength and a loss of corresion resistance. As reported by Dayton (9, p.460), the strongest alloy tested at 930°P creeps at a rate of 10-4 per cent per hour with a stress of only 4000 lb per sq in. Since a thin sheet is sandwiched between two layers of structural material, the loss of strength and corrosion resistance are secondary to the requirement that the fissionable material remain a solid. The coefficient of linear thermal expansion for 20.7 per cent zirconium by weight was not available; however, Dayton (9, p.libly) reported a coefficient for pure zirconium at 932°F for the "a" exis of 5.8 x 10-6 while that of pure uranium was anisotropic as shown by Saller (8, p.389). The coefficient of thermal expansion for this particular alloy was not found; however, from Saller (8, p.234) the composition for 20.7 per cent zirconium by weight indicated 3 phase changes for the temperature range up to 2200°F and whether the anisotropic effect of uranium would result in volume changes so severe as to make bonding or cladding impractical was not determined.

The low thermal conductivities of both zirconium and uranium are partially compensated for by low specific heats and the thin sheet effect of high area to volume ratio.

AND HE PARTY THAT I SHARE THE PARTY OF THE P

THE REAL ASSESSMENT OF THE PARTY OF THE PART THE PERSON NAMED IN COLUMN TWO IS NOT THE OWNER, NAMED IN COLUMN TWO IS NAMED IN COLUMN TWO Married Print 18 Secret white Designate and Ottolicy and Delivery TARREST AND ADDRESS OF NAME AND POST OFFICE ADDRESS OF THE PARTY AS NOT Designation for the party of the party of the last of the depart of planting the same of the last of second and AND VALUE DESCRIPTIONS OF THE PERSON NAMED IN COLUMN 2 IS NOT THE OWNER, the productions of the party of Makes tolerands among the secondary for something the states. THE RESIDENCE LABOUR PROPERTY AND PERSONS ASSESSED ASSESSED. the name "at" and was "They are not only a Charles not desirable than because the dispersation was not upon the new place. The a first Witness Asserted to Andrews our about any our nation of median power abtraction countries and suffer reduced to the age The sufference while may find and cuttingpoor and determine the the region of the contract of the last of the contract of the NAME AND ADDRESS OF PERSONS ASSESSED AND ADDRESS OF THE PARTY AND the Residence of the Part Street, Squared Stre

to produce the second control of the control of the

Uranium can be hot rolled without danger of cracking. Whether this particular alloy can be rolled into sheets with thicknesses as small as 0.03 inch was not determined.

The container material for temperatures up to 2200°F presents a considerable problem when restricted to present day material availability. The wrought cobalt-base alloys, such as Maynes Stellite-25, Allegheny-Ludlum Steel Corporation S-590 and S-816, and Westinghouse Electric Corporation Refractaloy 26 and 70, appear to have excellent corrosion resistance up to temperatures as high as 2000°F, Eichen and Jackson (10, p.556), in an exidizing atmosphere. However, Eichen and Jackson (10, p.539) reported the ultimate strength of Haynes Alloy 25 at 2200°F to be 4,000 lb per sq in. for a sheet 0.042 inch in thickness and annealed at 2200°F. Another major drawback may be the high residual activity in the form of gamma radiation which is characteristic of cobalt-base alloys.

Molybdenum is of interest as a high-temperature material because of its high melting point, 4750°F, high strength and stiffness at elevated temperatures. Northcott (11, p.106) reported creep-rupture strength for wrought are cast molybdenum, .45 per cent titanium, of 25,000 lb per sq in at 2000°F for 100 hours. The thermal expansion coefficient as reported by Northcott (11, p.26) ranged from 5.1 x 10⁻⁶ at 932°F to 7.2 x 10⁻⁶ at 3632°F. Thus the thermal expansion of both pure zirconium and molybdenum alloys appear to be very similar.

Charles to recommend to the same of the sa

The statement and a second color of the statement of the

However, alloying with uranium might alter the thermal expansion characteristic considerably.

Uncoated molybdenum is useless as a material of construction in oxidizing atmospheres above 1000°F. Norden (12, pp.298-30h) reported that so far no molybdenum-base alloys have been developed with satisfactory oxidation resistance; however, Hiester et al. (13) reviewed progress made with alloying techniques. Sherwood (1h, p.26) stated that molybdenum specimens protected by coatings 0.003 in thick and consisting primarily of molybdenum disilicide have successfully resisted oxidation at 3092°F for more than 1000 hours. However, the coating must be pore-free in order to prevent serious loss of material. Effects of the disilicide coating on heat transfer and thermal cycling were not included.

The high thermal conductivity of molybdenum, .32 cal per sec cm °C, and low specific heat, .061 cal per gm °C, permit it to be heated or cooled rapidly with low resultant stress which give it excellent resistance to thermal shock. The heat transfer characteristics are excellent due to the high thermal conductivity. The neutron moderation effects are small since it has a mass number of 96.

As brought out by Norden (12, p.298) a chief advantage of molybdenum over ceramics, cermets, and some super alloys is the variety of forms and sizes available on a production basis. By the arc-cast methods, sheets from 0.005 to 0.020 in. thick and 36 in. wide are available. Molybdenum has been joined

The Late of the La

waterfaces for facilities in an experience of manufacture forested

Applications of the property o

THE PARTY AND PA

successfully by spot welding, are welding, soldering, and brazing but precise controls must be used since excess heating will result in extremely brittle welds.

Conclusion

Molybdenum alloy, 0.45 per cent titanium, with a protective disilicide coating appeared to have the best probability of withstanding temperatures up to 2200°F in an oxidizing atmosphere besides being able to withstand thermal shock and offer near optimum heat-transfer characteristics. The inner fissionable phase of enriched uranium alloyed with 20.7 per cent by weight of zirconium, even though soft and pliable at this temperature range, which could be an advantage as far as the container material is concerned, should expel the heat produced internally due to the high surface to volume ratio. Whether or not the thermal expansion coefficient of the uranium-zirconium would be compatible with molybdenum and that a metallurgical bond would be formed with molybdenum was not ascertained.

the qualitative qualities are qualities tops or approximate which which we have all the property of the party of the party

ACCUPATION.

,

NUCLEAR ANALYSIS

The actual heterogeneous reactor configuration as shown in Figure 2 was treated as a homogeneous unreflected system using one-group diffusion theory to obtain the critical mass of uranium. The flattening of flux and reduction of critical mass by use of a reflector were not considered. The neutron flux necessary to produce a maximum fuel element temperature was calculated. The control mechanism necessary to maintain the flux level was not considered.

Development

The diffusion equation for thermal neutrons for steady state conditions as developed by Glasstone and Edlund (15, p.192) is

$$D \nabla^2 \beta (\overline{r}) - \Sigma_a \beta (\overline{r}) + pq (\overline{r}, \tau_{th}) = 0 . \qquad (20)$$

For an idealized fast reactor composed of atoms of material so heavy that neutrons upon colliding rebound without losing an appreciable amount of energy, the slowing down density term, pq (\bar{r}, τ_{th}) , may be considered negligible. If fast neutrons occur only at one energy, the source term can be written as

$$S = K \Sigma_{\mathbf{A}} \emptyset \qquad (21)$$

ALCOHOL MODEL CO.

THE PARTY OF THE P

THE RESIDENCE OF THE PARTY OF T

printed to make a second respective to the second of the second s

Using Glasstone and Edlund's (15) treatment the multiplication factor may be simplified to

$$K \approx v \left[\frac{\Sigma_f}{\Sigma_a}\right]_{\text{FUEL}} = \gamma$$
 (22)

by assuming that the fast fission factor, the resonance escape probability, and the fast neutron utilization are essentially unity. The fast neutron utilization term implies that the capture of neutrons by material other than the fissionable phase is negligible. Shapiro (16, p.130) states that the radiative capture cross sections for elements having mass numbers less than 100 and neutrons numbers less than 60 are not greater than 50 millibarns. If the fast neutron utilization factor were not unity, the treatment given by Kaplan (17, p.525) for thermal neutrons would still apply.

A combination of Equations 20 to 22 yields

$$\nabla^2 \beta (\overline{r}) + B^2 \beta (\overline{r}) = 0 \tag{23}$$

where
$$B^2 \equiv \Sigma_a (\gamma - 1)/D$$
. (24)

For a weak absorber as described by Glasstone and Edlund (15, pp.92,98), the diffusion coefficient is related to the transport mean free path by

THE PERSON NAMED IN COLUMN TWO IS NOT THE OWNER, THE PERSON NAMED IN COLUMN TWO IS NAMED IN C

MIDNEY 25 HT III THOUSE IN COLUMN 2

$$D = \frac{\lambda_t}{3} = \frac{1}{3\lambda_t} \tag{25}$$

where
$$\lambda_t \equiv \lambda_s (1 - \overline{\mu}_0)$$
 and $\overline{\mu} = 2/3A$. (26)

As a first approximation the macroscopic transport and scattering cross-sections are considered to be essentially equal except in cases where the actual values are known. The atomic masses of the uranium, zirconium, and molybdenum are high enough to make the $\overline{\mu}_0$ term of Equation 26 negligible. As a second approximation the elastic scattering cross section is expected to be equal roughly to the total cross section in the middle energy region of 1 to 500 kev where neutron gamma cross sections are small, Shapiro (16, p.78). The constants used in this analysis are listed in Appendix D with additional references.

Substitution of Equation 25 into 24 results in

$$B^2 = 3 \, \boldsymbol{\varepsilon}_{\mathrm{t}} \, \boldsymbol{\varepsilon}_{\mathrm{a}} (\boldsymbol{\eta} - 1) \tag{27}$$

The solution to Equation 23 for the buckling in terms of spatial dimensions of a right circular cylinder as developed by Glasstone and Edlund (15, p.213) is

$$B^2 = \left(\frac{2 \cdot 1105}{R}\right)^2 + \left(\frac{\pi}{H}\right)^2 \tag{28}$$

where R and H are the extrapolated radius and height of the cylinder in centimeters.

The absolute is at a substant or a substant or a substant or a substant of the substant of the

at attachment of the regions to recommend

to make all published out-out the delicant of materials and applicable in the state of the state

sent to and it are the summaries out out the track and the last of the application and tenters are the last

A combination of Equations 27 and 28 yields

$$3\boldsymbol{\varepsilon}_{t}\boldsymbol{\varepsilon}_{a}(\boldsymbol{\eta}-1) = \left(\frac{2\cdot 105}{R}\right)^{2} + \left(\frac{\pi}{H}\right)^{2} \tag{29}$$

Equation 29 is the approximate critical equation for an idealized fast homogeneous unreflected reactor in the form of a right circular cylinder. Appendix D shows the calculations that were used to determine the critical mass of uranium for a right circular cylinder with a physical diameter and length of 42 and 60 in, respectively.

The various component weights which were based on volume fractions with corresponding densities are calculated in Appendix D.

For the flux required to produce the maximum temperature change, the relationship

$$\beta_{\text{max}} = \frac{(T_{0.3} - T_{0.2}) G C_{p} \pi}{4E \Sigma_{f} t (L/D_{e})}$$
(30)

was used. The result is shown in Appendix D.

Discussion

The length of the heat exchanger was limited to 60 in. because Figure 7 showed that the maximum thrust per sq ft was obtained with an approximate L/D ratio of 120. This produced a 0.50 in. hydraulic diameter or a 0.25 in. fuel element spacing

AND DESIGNATION OF THE PARTY OF

the series of the comments of the contract of

The same of the sa

AND DESCRIPTION OF REAL PROPERTY OF THE REAL PROPERTY OF THE PARTY OF

NAME AND POST OFFICE ADDRESS OF TAXABLE PARTY.

AMERICAN AND PROPERTY.

per 21 to reside the process had not been a second and the second at the second at

The uranium-zirconium alloy with a composition of 79.3 per cent enriched uranium by weight was selected because of its 2,450°F melting point. A larger percentage of uranium would reduce this melting point which is considered a minimum for the high temperatures desired. The alloy composition was considered a constant and treated as a mixture for volume fractions since the actual density was unknown.

The component volume fractions of the heat exchanger resulted from trial and error thickness selections for the uranium-zirconium and molybdenum alloys which satisfied Equation 29 for a physical diameter and length equal to or less than 42 and 60 in, respectively. Since the physical length was fixed previously for maximum thrust per sq ft, the radius was considered variable.

The nuclear reactor, when treated as a homogeneous unreflected reactor according to Equation 29, was determined to
be critical at a physical diameter of 40 in and a length of
60 in. The large critical volume, 43.6 cu ft, was due to the
low macroscopic absorption of uranium, 0.0056 cm⁻¹, which resulted indirectly from the 75 per cent volume of air. A decrease in the volume fraction of uranium would decrease both
the total macroscopic absorption and transport cross sections
with a resultant increase in radius if the length remained
constant.

The weight of enriched, 90 per cent uranium required for the critical volume of 43.6 cu ft was 3,622 lbs. The amount The former between a consequence of makeur bedress and the consequence of the consequence

The same of the sa

And in case of the last to be the la

of zirconium and molybdenum was 960 and 3,470 lbs, respectively, which resulted in a total heat exchanger weight of 8,052 lbs. At current prices of approximately 30 dollars per gram, 50 million dollars would be required for the 90 per cent enriched uranium.

The flux that was necessary to generate a maximum wall temperature of 2200°F with a 0.0315 inch fuel element thickness was 8 x 10¹¹⁴ n per sq cm sec which assumed zero radial flux depression across the fuel element.

Since the radial flux will follow an approximate cosine distribution, it is apparent that for an average maximum wall temperature of 2000°F which was assumed to obtain 9,500 lbs of thrust with a 7.7 sq ft frontal area, the fuel elements near the centerline would operate at temperatures exceeding 2200°F. A proposed central control rod and reflector would tend to flatten the flux; however, the flux envelope resulting from these effects was not investigated.

The application of the one-energy group diffusion theory to this analysis was a realistic approximation because the transport mean free path, 8.7 cm, was small in comparison to the boundary distance. The extrapolation distance, R-r, is not generally equal to $0.71\,\lambda_{\rm t}$ for a fast reactor and its exact value can be obtained only by complicated methods, Glasstone (20, p.217). However, as pointed out by Glasstone (20, p.217), there is probably no point in a fast reactor that is more than two scattering or transport mean free paths from a

of characters and migraless on the second has, remarking in the plant of the plant in the plant

The product of the contract of

And the special of the second of the second

The formation of the control of the

boundary. This statement is supported by the large quantities of fuel required for the heat exchanger which has a minimum of 11 transport mean free paths in the radial direction.

An intermediate neutron energy reactor may offer the possibility of reducing the fuel quantity and weight while still maintaining a large volume fraction of air for compact heat transfer. The slowing down density term of Equation 20 was neglected because air, which composed 75 per cent of the heat exchanger volume, has a negligible moderation effect on neutrons. To continue this simplification the material for the fuel alloy and container was selected from elements with relatively high mass numbers. However, with the proper selection of lighter materials for the fuel alloy, container and reflector, the moderation effects may increase to the extent that this fast reactor would evolve into an intermediate reactor. It is believed that any trend toward a thermal reactor would decrease the fuel requirements for the configuration.

Conclusion

From a nuclear aspect this heat exchanger, which was composed by volume of 12.5 per cent alloyed fuel, 12.5 per cent molybdenum structure, and 75.0 per cent air, was critical at a diameter and length of 40 and 60 inches, respectively. The total weight was 8,052 lbs which included 3,622 lbs of enriched

Management and the later and t

90 per cent uranium. Approximately 50 million dollars would be required for this amount of uranium.

The flux that was necessary to generate a maximum wall temperature of 2200°F with a fuel element thickness, 0.0315 in. was reasonable at 1015n per sq cm sec.

With cost as a major consideration there is probably no point in a fast reactor more than several scattering or transport mean free paths from a boundary. An intermediate neutron energy reactor may offer the possibility of reducing the fuel quantity and weight for this configuration.

AND REAL PROPERTY AND PERSONS ASSESSED.

the fact of the party of the party of the state of

OVER-ALL PEASIBILITY CONCLUSIONS

For an air inlet velocity of M = 0.2, a weighted radial average of 2000°F for the maximum cylinder temperature, a frontal area of 7.7 sq ft, a 42 in diameter, and a 60 in length, the amount of heat transferred was substantial enough to produce 9,500 lbs of thrust with a flight Mach number of 2.5 at 37,500 ft.

In order to obtain an average temperature of 2000°F the fissionable material within the thin right circular cylinders had to reach a critical condition. This required 3,622 lbs of enriched 90 per cent uranium with an approximate cost of 50 million dollars. The heterogeneous nuclear heat exchanger was critical with a diameter and length of 40 and 60 in, respectively and a total weight of 8,052 lbs.

The low thrust to weight ratio may be compensated for by the negligible fuel burn up when compared to a turbojet system.

Molybdenum alloy, 0.45 per cent titanium, which was protested with a disilicide coating appeared to have the high temperature properties for the container material. Sandwiched between the molybdenum was an alloy of uranium-zirconium which contained 79.3 per cent uranium by weight. This binary system was selected because of the 2450°F melting point at the above composition. Whether a metallurgical bond could be formed between the molybdenum and fuel alloy was undetermined.

TOTAL CONTINUE OF A STREET

PRODUCTOR TO THE REAL PROPERTY OF THE PROPERTY OF THE PARTY OF THE PAR

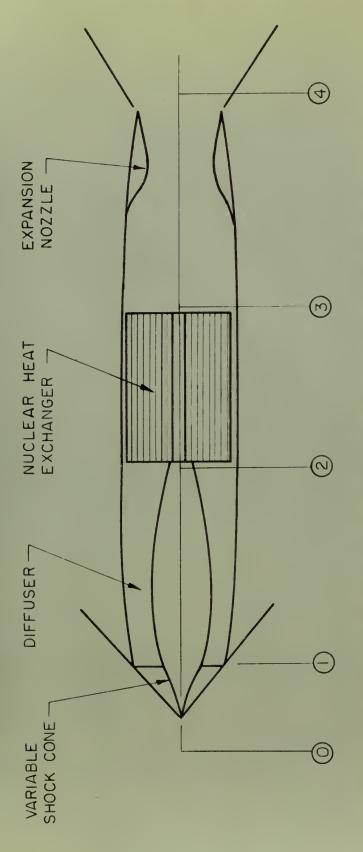
The second secon

From the above theoretical conclusions and limiting conditions, the over-all feasibility of a compact axial flow nuclear heat exchanger was favorable for thrust, metal availability, and nuclear criticality, borderline for thrust to weight ratio, and unfavorable for the cost of fuel. However, the proposed central control core and peripheral reflector should increase thrust and reduce the cost of fuel because of savings in critical mass.

An intermediate neutron energy reactor may offer the possibility of reducing the fuel requirements. With the proper selection of lighter materials for the fuel alloy, container, and reflector, the moderation effects may increase to the extent that this fast reactor would evolve into an intermediate reactor.

AND DESCRIPTION OF RESIDENCE AND PROPERTY OF PARTY AND PARTY OF PARTY OF PARTY AND PARTY OF PART

The state of the second second



temperature = 392°R, ambient pressure = 3.09 lb/in2. Assumed Idealized ramjet in flight at $M_0 = 2.5$, altitude = 37,500 ft, M = 0.2 at station 2. Figure 1.



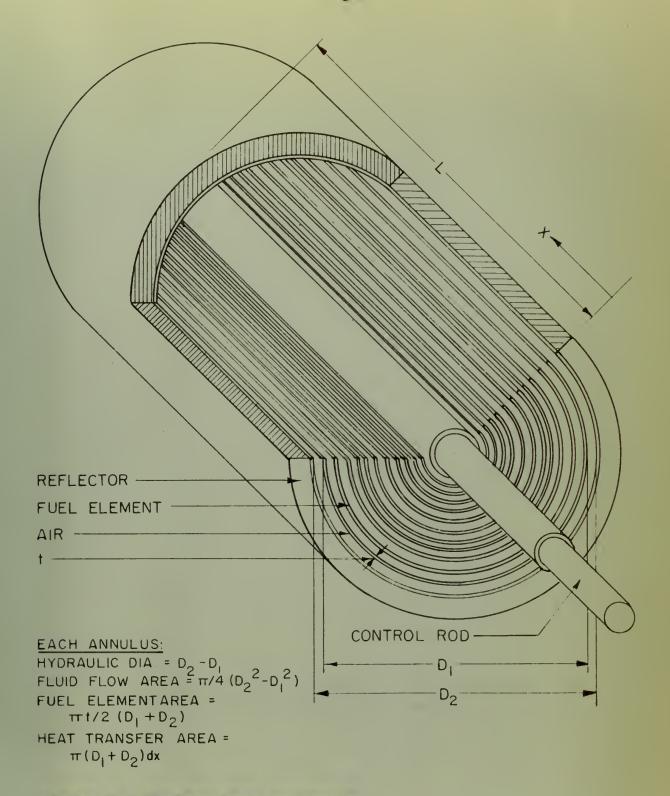


Figure 2. Nuclear heat exchanger.



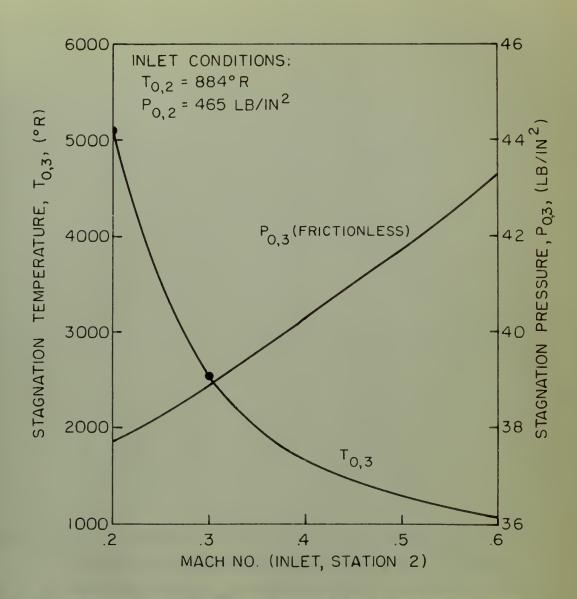


Figure 3. Thermal choking boundaries for a constant area frictionless chamber being heated.



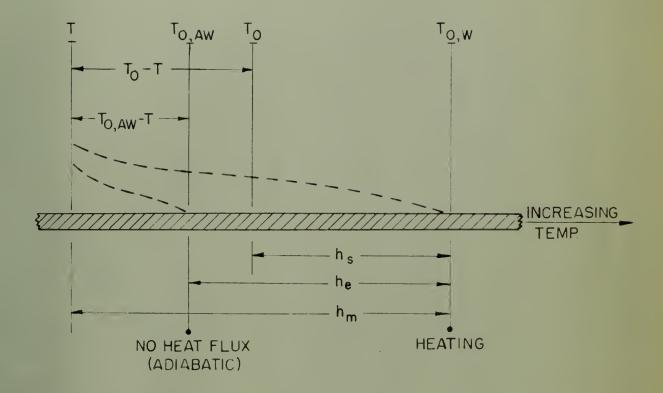
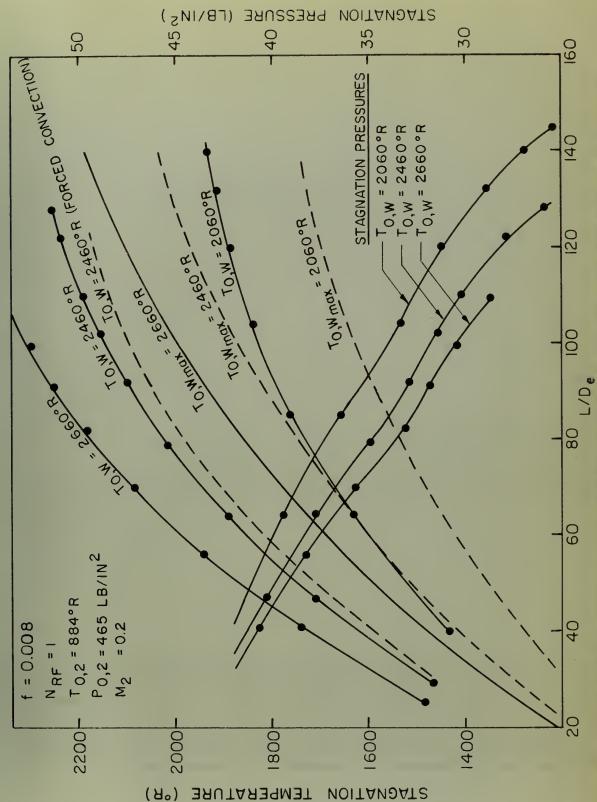


Figure 4. Relationship of various temperatures and heat transfer coefficients where $T_0 - T = V^2/2gJC_p$ and $N_{RF} = (T_{0,aw} - T)/(T_0 - T)$.





Curves of stagnation temperatures and pressures versus L/D ratios. Figure 5.



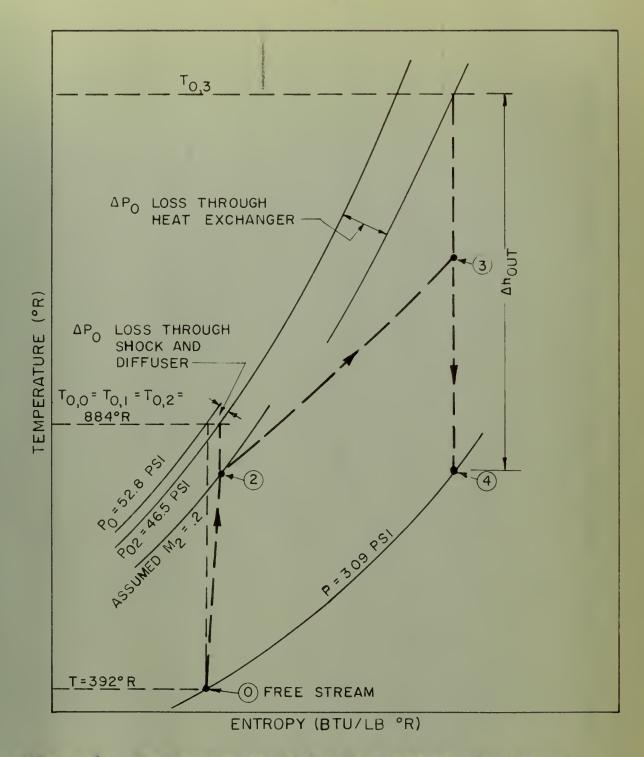


Figure 6. The Brayton cycle for temperature and entropy conditions for stations corresponding to Figure 1.

Values for dry air are from Reference (7).



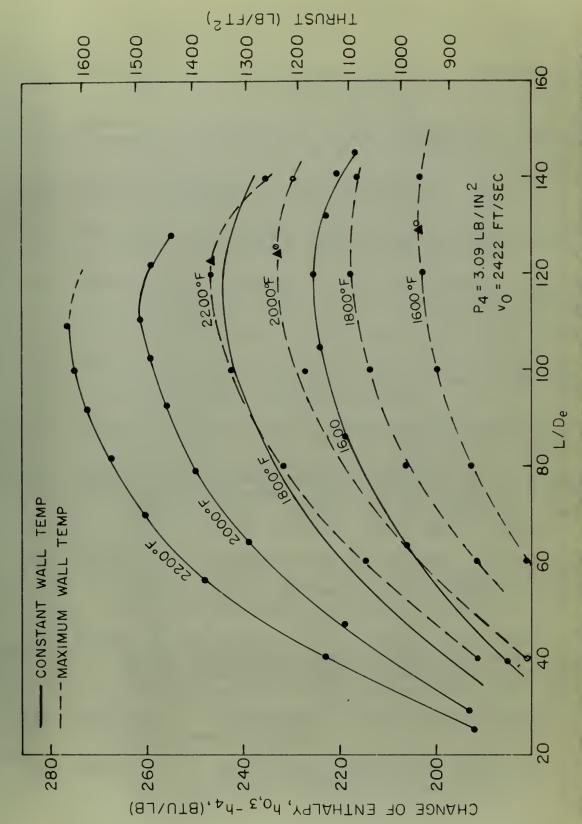
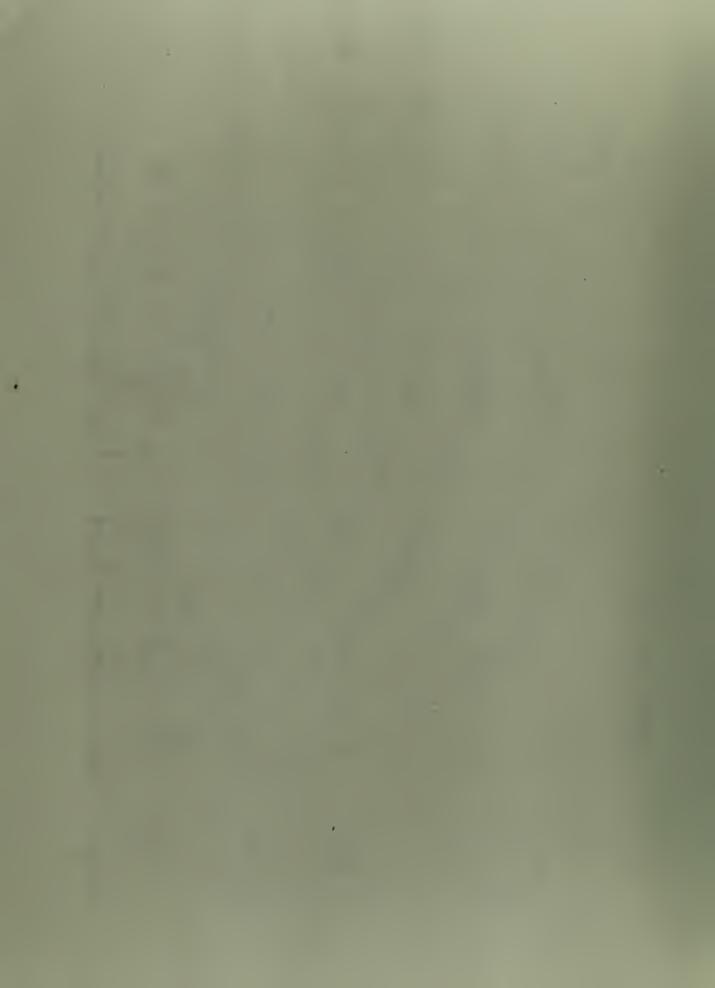


Figure 7. Change of enthalpy and thrust from exit of heat exchanger to exhaust.



LITERATURE CITED

- 1. Godsey, F. W., Jr. and Young, Lloyd A. Gas Turbines for Aircraft. N. Y., McGraw-Hill Book Co., Inc. (1949).
- 2. Bureau of Ordnance, U. S. Dept. Navy. Handbook of Supersonic Aerodynamics. Navord Report 1488 (Vol. 1). Washington, D. C., U. S. Govt. Print. Off. (1950).
- 3. McAdams, William H. Heat Transmission. 3rd ed. N. Y., McGraw-Hill Book Co., Inc. (1954).
- 4. Kays, W. M. and London, A. L. Compact Heat Exchangers. Palo Alto, Calif., National Press (1955).
- 5. McAdams, W. H., Nicolai, L. A., and Keenan, J. H. Measurements of Recovery Factors and Coefficients of Heat Transfer in a Tube for Subsonic Flow of Air. Trans. Am. Inst. Chem. Engrs. 42: 907-925. (1946).
- 6. Cope, W. P. The Friction and Heat Transmission Coefficients of Rough Pipes. Proc. Inst. Mech. Engrs. London. 145: 99-105. (1941).
- 7. Keenan, J. H. and Kaye, J. Thermodynamic Properties of Air. N. Y., John Wiley and Sons, Inc. (1945).
- 8. Saller, H. A. Uranium and Its Alloys. In U. S. Atomic Energy Commission. The Reactor Handbook. (AECD-3647). Vol. 3, Section 1. pp. 383-436. Washington, D. C., U. S. Govt. Print. Off. (1955).
- 9. Dayton, R. W. Zirconium and Its Alloys. In U. S. Atomic Energy Commission. The Reactor Handbook. (ARCD-3647). Vol. 3, Section 1. pp. 459-504. Washington, D. C., U. S. Govt. Print. Off. (1955).
- 10. Eichen, E. and Jackson, J. H. Cobalt Base Alloys. In U. S. Atomic Energy Commission. The Reactor Handbook. (AECD-3647) Vol. 3, Section 1. pp. 533-557. Washington, D. C., U. S. Govt. Print. Off. (1955).
- 11. Northcott, L. Molybdenum. N. Y., Academic Press, Inc. Publishers. (1956).
- 12. Norden, R. B. Molybdenum Combats High Temperature. Chem. Engr. 64, No. 2: 298-304. (Feb. 1957).

-12 -1 -3

- THE RESERVE OF THE PARTY OF THE
- The second control of the second of the seco
- The markets of the contract of the state of the contract of
 - The state of the state of the state of the state of
- The same of the sa

 - The state of the s
- where the state of the same of
- The same of the sa
 - The state of the same of the s
- The second secon

- 13. Hiester, N. K., Ferguson, F. A. and Fishman, N. High Temperature Technology. Chem. Engr. 64, No. 2: 237-252. (Feb. 1957).
- 14. Sherwood, E. M. Metals. In Campbell, I. E., Ed. High Temperature Technology. pp. 17-28. N. Y., John Wiley and Sons, Inc. (1956).
- 15. Glasstone, Samuel and Edlund, Milton G. The Elements of Nuclear Reactor Theory. Princeton, N. J., D. Van Nostrand Co., Inc. (1952).
- 16. Shapiro, Matthew M. Nuclear Physics. In U. S. Atomic Energy Commission. The Reactor Handbook. (AECD-3645) Vol. 1. pp. 61-364. Washington, D. C., U. S. Govt. Print. Off. (1955).
- 17. Kaplan, Irving. Nuclear Physics. Cambridge, Mass., Addison-Wesley Publishing Co., Inc. (1955).
- 18. Murray, Raymond L. Nuclear Reactor Physics. Englewood Cliffs, N. J., Prentice Hall, Inc. (1957).
- 19. Hughes, D. J. and Harvey, J. A. Neutron Cross Sections.
 U. S. Atomic Energy Commission. Brookhaven
 National Laboratory 325. Washington, D. C., U. S.
 Govt. Print. Off. (1955).
- 20. Glasstone, Samuel. Principles of Nuclear Reactor Engineering. Princeton, N. J., D. Van Nostrand Co., Inc. (1955).

- The other part of the control of the
- the second of th
 - Minimum and or world to describe the describe personnelly office of the contract of the contra
- A DESIGNATION OF THE PARTY OF T
 - the distance leading resident resident control profess off.
- THE RESERVE AND ADDRESS OF THE PARTY AND ADDRE
- and the same of th
 - Secured resident to reference about posterior and pro-

APPENDIX A

Stepwise Integration for Stagnation Temperature and Mach Number as a Function of L/D Ratio

The procedure used for integration of Equations 13 and 14 are outlined in Reference (2) and are reproduced in part using the following approximations:

- 1. The integral of dT_0/T_0 may be approximated by $2(T_{0,2} T_{0,1})/(T_{0,2} + T_{0,1})$.
- 2. The coefficients of dx/D_e in Equations 13 and 14 are taken at $M_{1.2} = (M_1 + M_2)/2$.

The inlet conditions are as follows:

$$M_1 = 0.2$$
 $N_{PR} = 0.664$
 $T_{0,1} = 884^{\circ}R$ $N_{RF} = 0.872$
 $P_{0,1} = 46.5 \text{ lb/in}^2$ $\lambda = 1.32$
 $T_{0,w} = 2460^{\circ}R$ $f = 0.008$

For a section 2 at a distance $x_2 - x_1$ downstream from section 1, the following computations are carried out.

a. Let
$$(x_2 - x_1)/D_0 = 30$$

b. Estimate
$$M_2 = 0.3$$
, $M_{1.2} = 0.25$

c. Calculate
$$2(T_{0,2} - T_{0,1})/(T_{0,2} + T_{0,1})$$
 from Equation 13.

more and the second section of the second section in

The state of the same of the s

AND ALL REAL PROPERTY AND ADDRESS OF THE PARTY AND ADDRESS OF THE PARTY

parallel at the annual control of

AND A WORLDOOD IN SUBMISSION OF SIX ASSESSMENT THE RESIDENCE AND ADDRESS OF THE PARTY ADDRESS OF

$$2\frac{(T_{0,2}-T_{0,1})}{T_{0,2}+T_{0,1}} = \begin{bmatrix} \frac{T_{0,W}}{(T_{0,2}+T_{0,1})/2}-1+\frac{(1-N_{EF})\frac{Y-1}{2}N_{1,2}}{1-\frac{Y-1}{2}N_{1,2}} \\ \frac{X_{1}}{D_{e}} \end{bmatrix}$$

$$2\frac{(T_{0,2}-884)}{T_{0,2}+884}=\frac{2460}{(T_{0,2}+884)/2}-1+\frac{(1-.872)(.2)(0625)}{1+.2(.0625)}.$$

2(1.32)(.008)(30)

$$T_{0.2} = 1640$$

d. Calculate $\rm M_2$ from Equation 15 and compare with assumed value of $\rm M_2$

$$M_2^2 - M_1^2 = \frac{.0625(1 + .2 \times .0625)}{(1 - .0625)} 2(.008)(30)$$

$$\left\{ (1 + 1.4 \times .0625)(1.32) \left[.95 \right] + 2 \times 1.4 \times .0625 \right\}$$

$$M_2^2 - M_1^2 = 0.0500$$
 $M_2^2 = 0.0500 + 0.0400$
 $M_2 = 0.3$ This agrees with the assumed value.

DESCRIPTION OF THE PARTY.

AND DESCRIPTION OF THE PARTY NAMED AND POST OFFICE ADDRESS OF TAXABLE PARTY.

Mary - "ye - "get

make a state of the

STATE OF THE PARTY OF THE PARTY OF THE PARTY.

- e. If step d was not successful, repeat the previous steps with an improved value of M2.
- f. After convergence is obtained, the process is repeated to find the Mach number at a section 3.

- American and America, Districtions from this A golds, 22 and and the control of t
- To settle employees to manufacture of a section to the team

APPENDIY B

Stepwise Integration for Stagnation Temperature and L/D Ratio as a Function of Mach Number

The procedures used for integration of Equations 14 and 17 are outlined in Reference (2). Equation 14 may be integrated approximately over a short interval of Mach number by assuming that the influence coefficients have a constant value corresponding to $(M_1 + M_2)/2$, that the magnitude of T_0 is constant during integration with a value of $(T_{0,1} + T_{0,2})/2$, and that the friction factor is constant. With these approximations and by the substitution of Equation 17 into 14, Equation 14 integrates into

$$M_2^2 - M_1^2 = 2 \left(\frac{T_{0,2}}{T_{0,1}} - 1 \right) \left[\frac{\overline{F}_{T_0}}{\frac{T_{0,2}}{T_{0,1}} + 1} + \frac{2\overline{F}_f}{\left(\frac{T_{0,W}}{T_{0,1}} \right) - \left(\frac{T_{0,2}}{T_{0,1}} \right) - 1} \right].$$
(17a)

For the same inlet conditions as were given in Appendix A, let

$$M_2 = 0.24$$
 $M_2^2 = 0.0576$
 $M_1 = 0.20$ $M_1^2 = 0.0400$
 $M_{1,2} = 0.22$ $M_2^2 - M_1^2 = 0.0176$

A STATE OF THE PARTY OF THE PAR

THE RESERVE THE PARTY OF THE PA

The property of the contract o

AN ADDRESS OF REAL PROPERTY AND ADDRESS OF THE PARTY AND ADDRESS OF THE PARTY.

nm.L

From Reference (2), \overline{F}_{T_0} = 0.05483 and \overline{F}_{T} = 0.00348. Equation 17a converged after three attempts to

$$\frac{T_{0,2}}{T_{0,1}} = 1.3 \mu \text{ or } T_{0,2} = 1188 \text{ er}$$

Equation 14 gave

$$\frac{x_2 - x_1}{D_e} = \frac{1}{2f} \ln \frac{\frac{T_{0,W} - 1}{T_{0,1}}}{\frac{T_{0,W} - \frac{T_{0,2}}{T_{0,1}}}{\frac{T_{0,1}}{T_{0,1}}} = \frac{1}{2(.008)} \ln \frac{2.785 - 1}{2.785 - 1.3hh} = 18.85$$

Since both Mach number and stagnation temperatures are known at each section, the following equations were used for computing the remaining properties of the air:

$$\frac{T_2}{T_1} = \frac{T_{0,2}}{T_{0,1}} \frac{\left(1 + \frac{y-1}{2} M_1^2\right)}{\left(1 + \frac{y-1}{2} M_2^2\right)} = 1.34$$

$$\frac{P_2}{P_1} = \frac{M_1}{M_2} \frac{T_2}{T_1} = 0.965$$

$$\frac{V_2}{V_1} = \frac{M_2}{M_1} = \frac{T_2}{T_1} = 1.392$$

The construct artist have been as To a modelline modelline and the second secon

TELL OF TAXABLE

THE REAL PROPERTY AND ADDRESS OF THE PARTY ADDRESS OF THE PARTY ADDRESS OF THE PARTY AND ADDRESS OF THE PARTY ADDR

$$\frac{P_{0,2}}{P_{0,1}} = \frac{P_2}{P_1} \frac{T_1}{T_2} = 0.720$$

$$\frac{P_{0,2}}{P_{0,1}} = \frac{P_2}{P_1} \left[\frac{1 + \frac{y-1}{2} M_2^2}{1 + \frac{y-1}{2} M_1^2} \right] = 0.965$$

Tables 1-3 which follow have calculate values for the various properties of air as a function of $L/D_{\rm e}$ ratios and different constant wall temperatures.

Table 1. Air properties as a function of Mach number for a constant wall temperature of 2660°R and a friction factor of .008

| М | T _O | L/De | T •R | P lb/in ² | V ft/sec | e lb/ft ³ | ^p ₀ 1b/in ² |
|------|----------------|--------|---------|-------------------------|-------------|-------------------------|--|
| 0.20 | 884 | 0 | 880 | 45.1 | 291 | 0.1385 | 46.5 |
| 0.24 | 1192 | 11.90 | 1182 | 43.6 | 405 | 0.1000 | 45.0 |
| 0.28 | 1484 | 25.50 | 1466 | 41.5 | 524 | 0.0770 | 43.0 |
| 0.32 | 1738 | 40.72 | 1709 | 39.2 | 646 | 0.0625 | 40.6 |
| 0.36 | 1939 | 56.04 | 1892 | 36.8 | 766 | 0.0528 | 38.1 |
| 0.40 | 2083 | 69.94 | 2020 | 34.3 | 881 | 0.0461 | 35.6 |
| 0.44 | 2182 | 81.74 | 2104 | 31.8 | 990 | 0.0410 | 33.0 |
| 0.48 | 2252 | 91.55 | 2158 | 29.6 | 1097 | 0.0372 | 31.8 |
| 0.52 | 2299 | 99.19 | 2182 | 27.3 | 1188 | 0.0340 | 30.4 |
| 0.60 | 2356 | 109.52 | 2182 | 23.6 | 1370 | 0.0294 | 28.6 |

described and the annual contract contract and the first and the analysis of the first and the first and the annual state of t

a war process and to maintain a management with all about

| | | | | 0 | |
|----------|-----------|-----------|--------|---|-------|
| | | | | | 26,6 |
| 0.00 | | 100 | STATE | | 15.00 |
| Barrier. | | 9400 | | | 10.0 |
| 0.00 | | | | | dis |
| | | | Steel | | Mes |
| | 20.00 | GUI | 46,55 | | 40 |
| 2+66 | 1000 | Bell. | | | 110 |
| 01400 | STORES. | | 9.70 | | Mai |
| | Diffe. | · (24/12) | 15,00 | | 23,0 |
| Aug I | distance. | | 17,101 | | 82.0 |

Table 2. Air properties as a function of Mach number for a constant wall temperature of 2460°R and a friction factor of .008

| M | T _O | L/D _e | r •R | p lb/in ² | V ft/sec | e 1b/ft ³ | P ₀ lb/in ² |
|------|----------------|------------------|---------|-------------------------|-------------|-------------------------|-----------------------------------|
| 0.20 | 884 | 0 | 880 | 45.1 | 291 | 0.1385 | 46.5 |
| 0.24 | 1184 | 13.43 | 1180 | 43.4 | 405 | 0.0998 | 44.9 |
| 0.28 | 1473 | 29.32 | 1462 | 41.5 | 527 | 0.0769 | 42.9 |
| 0.32 | 1711 | 46.70 | 1692 | 38.9 | 649 | 0.0625 | 40.3 |
| 0.36 | 1891 | 63.98 | 1863 | 36.4 | 768 | 0.0533 | 37.6 |
| 0.40 | 2018 | 79.33 | 1970 | 33.7 | 878 | 0.0465 | 34.8 |
| 0.44 | 2100 | 92.25 | 2040 | 31.2 | 985 | 0.0415 | 32.8 |
| 0.48 | 2155 | 102.6 | 2080 | 28.9 | 1085 | 0.0376 | 31.4 |
| 0.52 | 2192 | 110.7 | 2097 | 26.8 | 1182 | 0.0346 | 30.2 |
| 0.60 | 2236 | 121.9 | 2100 | 23.2 | 1363 | 0.0298 | 27.8 |
| 0.70 | 2258 | 128.2 | 2050 | 19.9 | 1590 | 0.0259 | 25.8 |

condition to the Parish or owners and the supramer of the car of the condition of the case of the case

| | Yelver . | | | | 2011 | 95 95 | |
|-------|----------|-----|---------|------|----------|----------|-------|
| 3.66 | 2000 | | 1430 | | 6.7 | - | 1545 |
| 7-10 | BASOWI | 201 | 2,00 | | 4.4 | | |
| TABLE | 100 | | Polici. | 146 | Herr | | 1550 |
| 5.00 | BUILD | | Bytte | | 91-10 | | 100 |
| 6-75 | | | | | 53,170 | | BEVE |
| 1.46 | 1000 | | 1700 | | | | 10/40 |
| HARE | Secretar | | 5400 | | | | |
| 2242 | ETP/A | | | | Jugan | | 0/-7 |
| 4200 | Albiye | | Balla | | | | 12.00 |
| | | | | | - Wellie | | 130 |
| | | | Total | 1000 | Serv. | | 11/4 |

Table 3. Air properties as a function of Mach number for a constant wall temperature of 2060°R and a friction factor of .008

| M | To | L/De | T | P | Po |
|------|------|-------|------|--------------------|--------------------|
| | •R | | •R | lb/in ² | lb/in ² |
| .20 | 884 | 0 | 880 | 45.1 | 46.5 |
| .24 | 1178 | 17.95 | 1160 | 43.4 | 44.6 |
| .28 | 1437 | 39.75 | 1411 | 41.0 | 42.2 |
| .32 | 1631 | 63.55 | 1592 | 38.2 | 39.4 |
| .36 | 1760 | 85.95 | 1711 | 35.2 | 36.3 |
| .40 | 1839 | 104.7 | 1775 | 32.3 | 33.3 |
| · 44 | 1886 | 102.0 | 1811 | 29.6 | 31.1 |
| .48 | 1914 | 131.9 | 1830 | 26.9 | 28.8 |
| .52 | 1935 | 140.6 | 1830 | 24.3 | 26.9 |
| .60 | 1945 | 145.6 | 1800 | 21.1 | 25.3 |

makes a but official to make a part of the part of the

| | | 1 | | 2 | |
|---------|-------|---------|--------|-------|-------|
| | 2,462 | | 6 | 410 | 154 |
| | 60 | | 10000 | | 150 |
| Section | 142 | | | TOP - | 1776 |
| | Selle | | 150+01 | | (434) |
| | 400 | | | | de |
| + B-SE | 100 | | | | 1534 |
| 1.17 | 1470 | APRIL . | 0,000 | | ALL |
| B-IC | 7-79 | | 1400 | | |
| 7-10 | | | double | | 150 |
| | G-17 | | | No. | 1774 |

APPENDIX C

Calculation for Outlet Stagnation Temperatures as a Function of L/D_e Ratios and Maximum Wall Temperatures

With reference to Figure 2, for a heat balance on the fuel cylinders without axial conduction requires that

$$T_{0,w} - T_{0} = \frac{Q_{max} t}{2h} \sin \frac{\pi x}{L}$$
 (A)

The longitudinal volumetric heat source term is zero when x = 0, or L, which assumes zero extrapolation distance.

Neglecting axial conduction within the fluid and any heat losses, a heat balance on the air in the annuli requires that

$$T_{O,W} - T_{O} = \frac{e^{VC_{D}D_{e}}}{lth} \frac{dT_{O}}{dx} .$$
 (B)

Equating Equations A and B and performing the required integration results in

$$T_0 - T_{0,2} = \frac{2Q_{\text{max}} t}{G C_p \pi} \left(\frac{L}{D_e}\right) \cdot \left[1 - \cos\frac{\pi x}{L}\right]. \quad (C)$$

Adding Equations A and C to eliminate the stagnation temperature of the air yields

as the newspaper of the purpose of t

And resident and a leaf of resident at an experience of the

the contract of the last of th

And other measurement and it and the property and the property of

word and the special part of contrasts of a plan is an additional position.

Salating with cold for department.

$$T_{0,w} - T_{0,2} = \frac{Q_{\text{max}} t}{2h} \sin \frac{\pi x}{L} + \frac{2Q_{\text{max}} t}{\pi G C_p} \left(\frac{L}{D_e}\right) \left[1 - \cos \frac{\pi x}{L}\right] . \quad (D)$$

Substitution of Equation 7 with $\lambda = 1$ into Equation D results in

$$T_{0,w} - T_{0,2} = \frac{Q_{\text{max}} t}{G C_{p}}.$$

$$\left[\frac{1}{f} \sin \frac{\pi x}{L} + \frac{2}{\pi} \left(\frac{L}{D_{e}}\right) - \frac{2}{\pi} \left(\frac{L}{D_{e}}\right) \cos \frac{\pi x}{L}\right].$$
(E)

From Equation E there appears to be a maximum which represents the highest temperature of the fuel cylinder in the longitudinal direction. The point at which the maximum is attained is found by differentiating Equation E and setting the result equal to zero.

$$x_{\text{max}} = \frac{L}{\pi}$$
 are $\tan \left(-\frac{\pi}{2\Gamma} \frac{D_e}{L}\right)$ (F)

The maximum wall temperature occurring in the fuel cylinder can now be obtained by inserting Equation F into equation E. For purposes of computation, it is convenient to represent $\pi x/L$ by the symbol a.

editions of an internal paper 2 of X again Y and the an executive against a second based on the second bas

THE RESIDENCE OF THE RESIDENCE OF THE PARTY OF THE PARTY

THE PARTY AND REAL PROPERTY OF THE PARTY OF

By substituting G into Equation F, results in

$$\tan a_{\max} = -\frac{\pi}{2f} \frac{D_e}{L} \tag{H}$$

Utilizing Equation H. Equation E reduces to

$$T_{0,w_{\text{max}}} - T_{0,2} = \frac{2Q_{\text{max}} t}{G C_p \pi} \left(\frac{L}{D_e}\right) \left[1 - \sec \alpha_{\text{max}}\right]$$
 (1)

From Equation 17, which was previously derived,

$$\frac{dT_0}{T_{0,w}-T_0} = 2f \frac{dx}{D_e}$$

and introducing Equation A, which was simplified by substituting $h = \frac{GG_p}{2}$, the following differential equation was derived:

$$dT_0 = \frac{2Q_{\text{max}} t}{G C_p D_0} \quad \sin \frac{\pi x}{L} dx \tag{J}$$

Integrating Equation J results in

$$T_{0,3} - T_{0,2} = \frac{\mu Q_{\text{max}} t}{G C_p \pi} \left(\frac{L}{D_e}\right)$$
 (K)

However, Equation K may be combined with Equation I to eliminate Q_{max} t, which results in

of which the party of the later of

Delineas something of the same of the same and

characteristics and married married married

and decreased on markets on many of markets of markets and the first of markets and the first of the first of

IN PERSONS ASSESSED.

AND THE PARTY OF T

$$T_{0,3} - T_{0,2} = 2 \frac{T_{0,w_{max}} - T_{0,2}}{1 - \sec \alpha_{max}}$$
 (18)

The magnitude of $T_{0,w_{\rm max}}$ utilized the previously selected wall temperatures. Since $T_{0,2}$ was the given inlet conditions, $a_{\rm max}$ was computed for various $L/D_{\rm e}$ ratios resulting in $T_{0,3}$ as a function of $T_{0,w_{\rm max}}$ and $a_{\rm max}$. Table 1 of Appendix C presents calculated values of $T_{0,3}$ which are shown in Figure 5.

Table 4. Outlet stagnation temperatures as a function L/De ratios and maximum wall temperatures

| L/D _e | | 40 | 60 | 80 | 100 | 120 | 140 |
|------------------|----|--------|-----------|-----------------------|-------------|--------|--------|
| tana | | -4.910 | -3.270 | -2.450 | -1.965 | -1.635 | -1.405 |
| secamax | | -5.010 | -3.420 | -2.650 | -2.210 | -1.920 | -1.728 |
| 1-seca max | | 6.010 | 4.420 | 3.650 | 3.210 | 2.920 | 2.728 |
| | | To,2 = | 884°R and | d To, w _{ma} | = 2660 | o-R | |
| To,3 | oR | | 1688 | | | 2101 | 2184 |
| | | To,2 = | 884°R and | d To, w _{ma} | = 2460 | o R | |
| To,3 | •R | 1408 | 1597 | 1748 | 1865 | 1964 | 2038 |
| | | To,2 = | 884°R and | d To, w _{ma} | = 2060 x | oR | |
| To,3 | oR | 1275 | 1415 | 1529 | 1616 | 1689 | 1745 |

The second section of the second section of

plus makes the bit have been been been and the state of

| | 117414 | | | TEAT- | |
|---------|--------|-------|--------|---------|--------|
| | apple | - | Title- | BANK I- | 167.7- |
| numed . | 9/04 | 17793 | | | 250 |
| | | 4000 | | | |
| | | | | | |
| | - 20 | | | | |
| 6,67 | | | | | |
| | | 200 | | | |
| | | | | | |

APPENDIX D

Critical Calculations for a Homogeneous Unreflected Reactor

The effective neutron energy was taken as 0.3 Mev based on flux spectra computed for existing fast reactors by Murray (18, p.69). From the same Reference the following constants were used:

$$\sigma_{\rm t} = 0.18 \, \rm barns$$
 $\sigma_{\rm f} = 0.0$ $\sigma_{\rm t} = 7.4 \, \rm barns$

$$\sigma_{\rm c} = 1.78 \text{ barns}$$
 $\sigma_{\rm f} = 1.50 \text{ barns}$ $\sigma_{\rm t} = 7.4 \text{ barns}$

From Hughes and Harvey (19) at 0.3 Mev, the total cross sections of the selected reactor materials and air were as follows:

$$Zr$$
: $\sigma_{T} = 9.0 \text{ barns}$

Mo:
$$\sigma_p = 9.0$$
 barns

$$N: \sigma_m = 4.0 \text{ barns}$$

0 :
$$\sigma_{\rm T} = 3.0 \; {\rm barns}$$

The above total cross sections were considered to be equal approximately to the scattering or transport cross sections.

A 90 per cent enriched uranium mixture by weight produced

$$\frac{235 \text{ N}_5}{238 \text{ N}_8 + 235 \text{ N}_5} = .90 \qquad \text{or} \qquad \frac{\text{N}_8}{\text{N}_5} = 0.105 \quad .$$

THE RESERVE AND PERSONS NAMED IN COLUMN 2 IN COLUMN 2

AND THE RESIDENCE OF THE PARTY OF THE PARTY

AND RESIDENCE AND ADDRESS OF THE PARTY OF TH

AND ADDRESS OF THE PARTY OF THE

The solution to Equation 22 with the above result yielded

$$\eta = \frac{\sigma_{f,5}}{\sigma_{a,5} + \sigma_{a,8}} (N_8/N_5) = 2.51 \frac{1.50}{1.78 + (.18)(1.05)} = 2.09$$

Since the alloyed density of the uranium-zirconium system was unknown, the alloy, 79.3 per cent uranium by weight, was treated as a mixture to obtain the following volume fractions:

$$f_{238} = 0.055$$
 $f_{235} = 0.512$
 $f_{zr} = 0.432$

A hydraulic diameter, $D_e = 0.5$ in, was obtained with a selected $L/D_e = 120$ and L = 60 in. The following thicknesses were assumed:

Uranium-zirconium alloy: 0.03125 in

Double wall of molybdenum: 0.03125 in

Hydraulic diameter/2: 0.25000 in

Thickness of air space: 0.18750 in

The approximate component volume fractions obtained from the above solidity factors were as follows:

$$f_N = 0.5925$$
 $f_0 = 0.1575$
 $f_{238} = 0.0070$
 $f_{235} = 0.0640$
 $f_{Zr} = 0.0540$
 $f_{Mo} = 0.1250$

THE RESIDENCE OF THE PARTY OF T

A SECOND ROLL OF THE PARTY OF A P

AND REAL PROPERTY AND PERSONS ASSESSED ASSESSED ASSESSED.

The macroscopic cross sections were computed for each of the above volume fractions which gave for a total

$$\Sigma_{t} = 0.1183 \text{ cm}^{-1}$$
 and $\Sigma_{a} = 0.0056 \text{ cm}^{-1}$.

Solution to Equation 27 yielded

$$B^2 = 21.7 \times 10^{-4} \text{ om}^{-2}$$
.

With a physical length equal to 60 in or 152.4 cm and the transport mean free path equal to 8.45 cm, the length including the extrapolation distance was

$$H = h + .71 \lambda_t = 152.4 + 6.0 = 158.4 \text{ om}$$
.

The solution to Equation 29 for the critical radius including the extrapolation distance was 56.7 cm; the physical radius was 50.7 cm or 20 in.

Since the total volume was 12.35×10^5 cu cm, the following component weights were determined by using the previously computed volume fractions with corresponding densities:

To compare the contract of the

the same of the state of the

American product of patients.

1 19 May - Call - 30

MARK A PROPERTY THREE AREAS AS NO DE ON TABLE ON THE SECOND THREE PARTY OF THE SECOND THREE PARTY OF THE SECOND THREE PARTY ON THE SECOND THREE PARTY OF THE SECOND THREE PARTY ON THE SECOND THREE PARTY OF THE SECOND THREE PARTY OF THREE PARTY

The contract of the Park Contract of the Contr

Contract and the first beautiful and makes the second and the

AND THE REAL PROPERTY AND ADDRESS OF THE PARTY AND ADDRESS OF THE PARTY

From Equation K, Appendix C, the flux that was necessary to generate a maximum wall temperature of 2200°F with a fuel element thickness, 0.0315 in, was obtained with

$$Q_{\text{max}} = \frac{(T_{0.3} - T_{0.2}) G G_p \pi}{\text{ME } \Sigma_f t (L/D_e)}$$

where

$$E = 200 \text{ MeV/fission}$$

$$E_f = 0.045 \text{ cm}$$

$$MeV = 1.52 \times 10^{-16} \text{ BTU}$$

$$G = 40.4 \text{ lb/ft}^2 \text{ sec}$$

$$C_p = .2671 \text{ BTU/lb °P}$$

$$L/D_0 = 120$$

$$T_{0,3} - T_{0,2} = 2101 - 884° \text{ from Figure 5.}$$

The flux obtained from the above equation was $8.25 \times 10^{14} n/$ cm² sec.

Name of Street, or other Persons of the Street, or other persons or other

HE SHEWARDS

m date - 2

THE RESIDENCE AND

- Person named at 1971

The state of the season

BEE - 1974

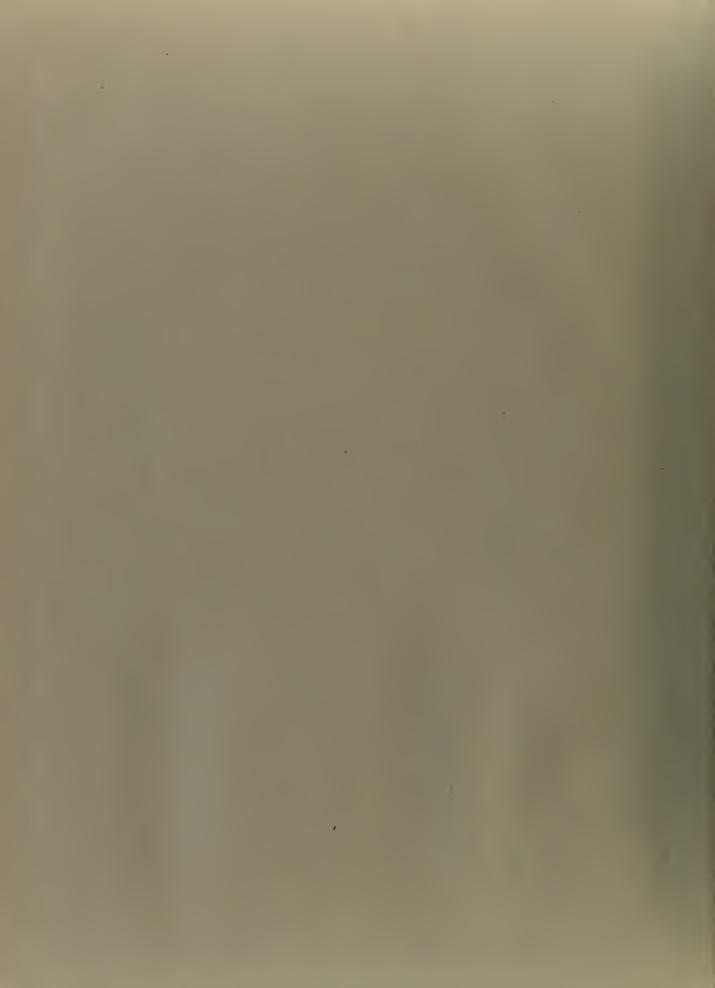
- - HE SHIP THE SHIP - HOLE THE STATE OF

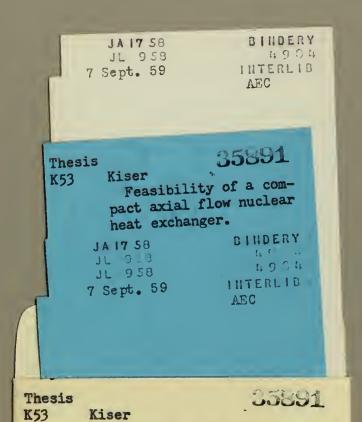
William State and make the same and and a second district the same and











Feasibility of a compact axial

flow nuclear heat exchanger,



HINGE THE PROPERTY OF THE PROP